

HYPOIODITE REACTION OF N-ACETYLJERVINE:
AN UNPRECEDENTED RADICAL ADDITION TO A CARBONYL GENERATED
BY β -CLEAVAGE OF ALKOXYL RADICAL (1)

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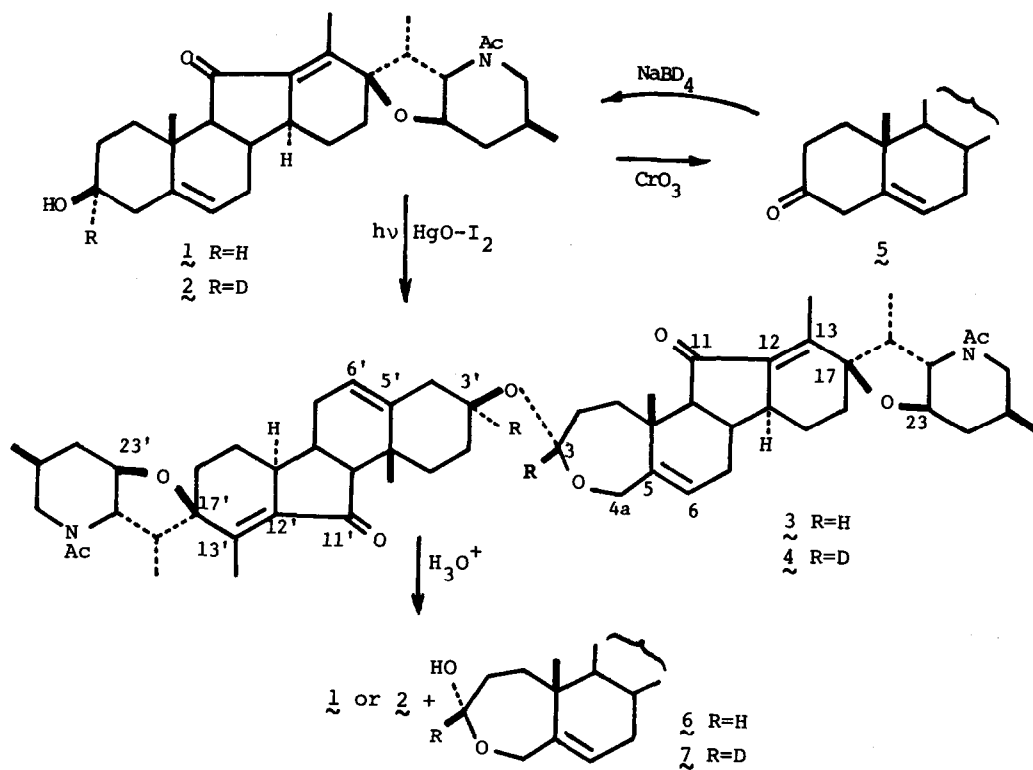
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Our recent investigation disclosed that the products of the hypiodite reaction of steroids with 3-hydroxy- Δ^5 system are 4-iodo-3,4-seco-steroid-5-en-3-als and oxidomethylene-3 α ,5 α - Δ -norsteroids. (2)

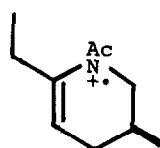
We wish to report on the results of hypiodite reaction of \underline{C} -nor- \underline{D} -homo-steroid with 3-hydroxy- Δ^5 system as the part structure. An unexpected variation between steroids and \underline{C} -nor- \underline{D} -homosteroids with regard to the type of the products was found.

Thus, the irradiation of N-acetyljervine 1 (3) (800 mg) in benzene (60 ml) in the presence of HgO and I₂ (each 3 equivalents) through pyrex for 12 hr by a 100-W high pressure Hg arc lamp under an argon atmosphere afforded a mixture of products separated by the preparative t.l.c. to afford five new compounds. The structure of a major product 3, m.p. 216-218°, (25% yield) is described in the present paper, and the structures of the remainder and the probable sequences of the formations of all five products are discussed in the following paper.

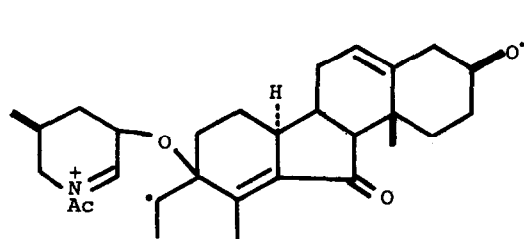
The molecular formula of 3 was found to be C₅₈H₈₀O₈N₂ by elemental analysis and the molecular weight determination by the osmometric method. (4) (Found; 918 Calcd for C₅₈H₈₀O₈N₂, 933.28). The i.r., u.v. and ¹H n.m.r. spectra of 3 confirmed that the C, D and the heterocyclic rings of 1 are not altered by the photolytic procedure. Thus, the presence of the α,β -unsaturated carbonyl group in 3 was confirmed by the i.r. spectrum (Nujol, 1633 and 1713 cm⁻¹) and by the u.v. spectrum ($\lambda_{\text{max}}^{\text{CHCl}_3}$ 253 nm, ϵ 31,100). In the ¹H n.m.r. spectrum (100 MHz, CDCl₃), the N-acetyl, the 18- and the 19-methyl groups appeared as the three 6-



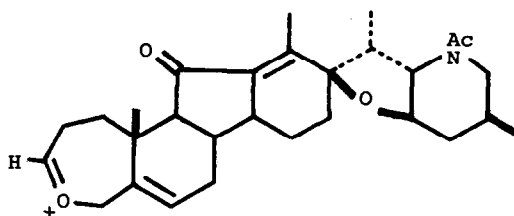
Scheme



8 m/e 167 (100%)



9 m/e 465 (2%)



10 m/e 466 (1%)

proton singlets at τ 7.89 at τ 7.73 and at τ 8.98, as well as the 21- and the 26-methyl groups as the two 6-proton doublets at τ 8.95 ($J=9.0$ Hz) and at τ 9.11 ($J=8.3$ Hz).

The clue for the structure of **3** was given by the presence of very characteristic four one-proton signals at τ 4.39 (diffused doublet, $J=4.0$ Hz), at τ 4.63 (diffused doublet, $J=4.0$ Hz), at τ 5.10 (triplet, $J=5.8$ Hz) and at τ 5.65 (doublet, $J=13.0$ Hz) in the ^1H n.m.r. spectrum.

Decoupling experiments showed that two signals at τ 4.39 and at τ 4.63 collapsed to two clear singlets on double irradiation at τ 7.5 \pm 0.1 ppm while a triplet at τ 5.10 was only partially decoupled. The properties of the two signals at τ 4.39 and at τ 4.63 in this double irradiation is reminiscent of those of C-6 olefinic proton of steroids and suggests the presence of a couple of protons, both of which are in a similar environment to C-6-olefinic proton of steroids. The diffused triplet at τ 5.10 was also partially decoupled to broad singlet ($W_H=7.0$ Hz) on irradiation at τ 8.3. This proton which is responsible for this triplet was confirmed to be derived from the C-3 α -H of N-acetyljervine since this signal is absent in the spectrum of the product **4**, m.p. 220-222 $^\circ$ obtained from 3 α -deuterio-N-acetyljervine **2**, m.p. 224.5-226.5 $^\circ$ prepared by the Jones oxidation of N-acetyljervine to N-acetyljerva-5,12(13)-diene-3,11-dione **5**, m.p. 172.5-174.5 $^\circ$, followed by NaBD_4 reduction.

The chemical shift and double irradiation also confirmed that the doublet at τ 5.65 should be attributable to geminal coupling of isolated oxygen-bearing methylene protons; on irradiation of the centre of a doublet at τ 6.46 ($J=13.0$ Hz) this doublet collapsed to a singlet, and the irradiation at τ 5.65 caused a collapse of a doublet at τ 6.46 without alteration of the signals at τ 4.39, at τ 4.63 and at τ 5.10.

Hydrolysis of either **3** or **4** in THF with dilute aqueous HCl at room temperature affords a couple of alcohols, N-acetyljervine and amorphous alcohol **6** or 3 β -deuterio-N-acetyljervine and deuterated **7**.

The ^1H n.m.r. spectrum (100 MHz, CDCl_3) of **6** revealed that, besides the signals due to the N-acetyl, the 18-, 19-, 21- and 26-methyl groups, a series of one-proton signals at τ 4.41 (diffused doublet, $J=3.9$ Hz), at τ 4.86 (triplet, $J=$

6.1 Hz) and at τ 5.59 (doublet $J=13.3$ Hz), which showed remarkable similarity to the signals of 3 in the coupling patterns, the chemical shifts and behaviours for the decoupling of each signals. However, the spectrum of 6 lacked a signal corresponding to a diffused signal at τ 4.63 in 3. The i.r. and u.v. spectra of 6 indicated the presence of α,β -unsaturated ketone function of jervine type. These coupled with the formation of N-acetyljervine in this hydrolysis, prove the signal at τ 4.63 in 3 is due to the C-6-H of N-acetyljervine unit in the compound 3.

All results on 3 above can now be reconciled with perfect consistency in terms of the dimeric structure formulated in the Scheme.

The mass spectrum of 3 revealed no M^+ peak and the highest fragments at 465 (M^+-467) and at 466 with the base peak at m/e 167. These fragment peaks are assignable to species 8, 9 and 10. The ^{13}CFT n.m.r. and ^1H off resonance spectra of 3 (CDCl_3 , 25.1 MHz, ppm from TMS) fully supported the assigned formula 3. Except C-17, C-17', N-Ac, N'Ac and C-23, C-23' which appeared respectively as the coincident singlets at 172.97, 85.42 and 72.68, all the resonances due to SP^2 carbon atoms and oxygen-bearing carbon atoms appeared as a pair of signals. The assignments of these signals are as follows; C-11-CO and C-11'CO (206.28 and 206.65), C-5 and C-5' (145.79 and 146.28), C-13 and C-13' (142.46 and 144.22), C-6 and C-6' (120.67 and 125.95), C-3 of oxepane ring (99.08), C-3' (75.72).

The probable sequence of the formation of 3 together with those of other products will be the subject of an accompanying Communication, but it is noted that the present reaction is a rare example in which radicals combine with a polar π bond generated by β -cleavage of an alkoxyl radical.

REFERENCES AND NOTES

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- (4) Hitachi 115 instrument was used, using benzene as the solvent.

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